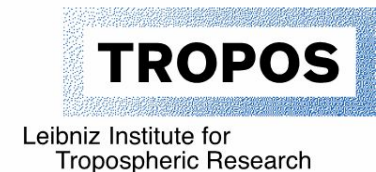


Atmospheric Aerosol Physics, Physical Measurements, and Sampling

Particle Diameters & Size Distribution

São Paulo School of Advanced Science on Atmospheric Aerosols:
properties, measurements, modeling, and effects on climate and health



Mobility or Stokes Diameter

Particle Diameter

The particle size is defined either by the diameter, D_p

The diameter is typically given in μm [10^{-6} m] or nm [10^{-9} m]

The diameter is normally defined as equivalent diameter (non-spherical particles are described as spheres).

Particle diameter definitions

Different measurement principals result in different definitions for the particle diameter:

- Stokes (mobility) diameter, $D_{p,St}$
- Optical diameter, $D_{p,Opt}$
- Aerodynamic diameter, $D_{p,Ae}$

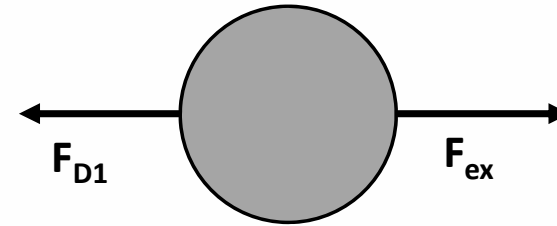
Important is also the volume equivalent diameter, $D_{p,Ve}$

Stokes Diameter (Mobility Diameter)

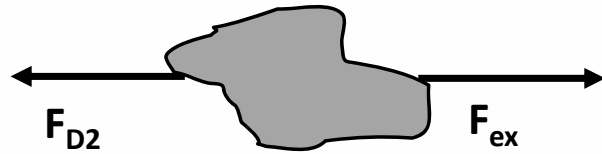
The **Stokes diameter** is defined for a **uniform particle motion**, where the external force equals the drag force. The motion is independent of the particle density.

$$\vec{u}_p = \text{constant}$$

$$\vec{F}_{ex} = \vec{F}_D = \frac{3\pi \cdot \eta \cdot \vec{u}_p \cdot D_p}{C_C}$$



- For a spherical particle, the Stokes diameter $D_{p1,St}$ can be determined then by the drag force and the particle velocity.
- In case of a spherical particle, the Stokes diameter is equal the geometric diameter and the volume equivalent diameter $D_{p1,Ve}$.

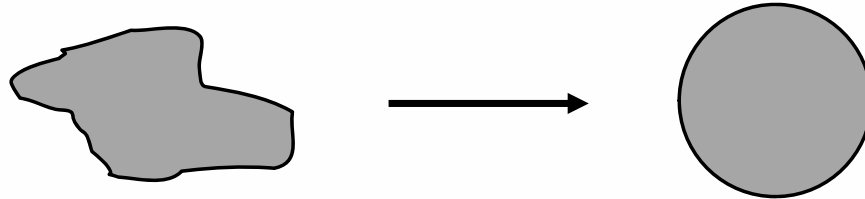


For an irregular particle, the Stokes diameter, if $D_{P2,St} = D_{P1,St}$

$$\vec{u}_{P2} / \vec{F}_{D2} = B_2 = B_1 = \vec{u}_{P1} / \vec{F}_{D1}$$

The volume equivalent diameter of an irregular particle can be calculated by knowing additionally the dynamic shape factor.

$$\vec{F}_D = \frac{3\pi \cdot \eta \cdot \vec{u}_p \cdot D_{P,Ve}}{C_C} \cdot \chi$$

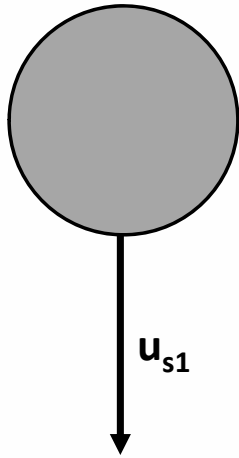


The volume equivalent diameter $D_{P2,Ve} < D_{P1,Ve}$

Aerodynamic Diameter

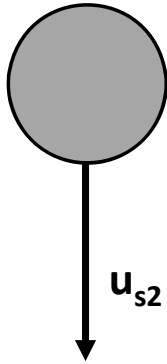
Aerodynamic Diameter

The aerodynamic particle diameter is used when the drag force depends on the particle density such as for sedimentation velocity.



For a certain sedimentation velocity, the aerodynamic particle diameter can be calculated by using $\rho_p = \rho_0 = 1$

$$\vec{u}_s = \frac{\rho_p \cdot D_p^2 \cdot C_C \cdot \vec{g}}{18\eta} = \frac{\rho_0 \cdot D_{p,Ae}^2 \cdot C_C \cdot \vec{g}}{18\eta}$$

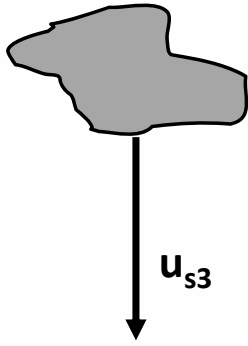


If the particle density is $\rho_p \neq 1$ and the sedimentation velocities are $\vec{u}_{s2} = \vec{u}_{s1}$ the aerodynamic diameters are identical $D_{P2,Ae} = D_{P1,Ae}$

The Stokes diameter can be calculated from the aerodynamic diameter, if the particle density is known.

For a spherical particle with the density $\rho_p > 1$, the Stokes (volume equivalent) diameter is smaller than the aerodynamic particle diameter and can be calculated to:

$$D_{P2,St} = D_{P2,Ve} = D_{P1,Ae} \sqrt{\frac{\rho_0}{\rho_P}}$$



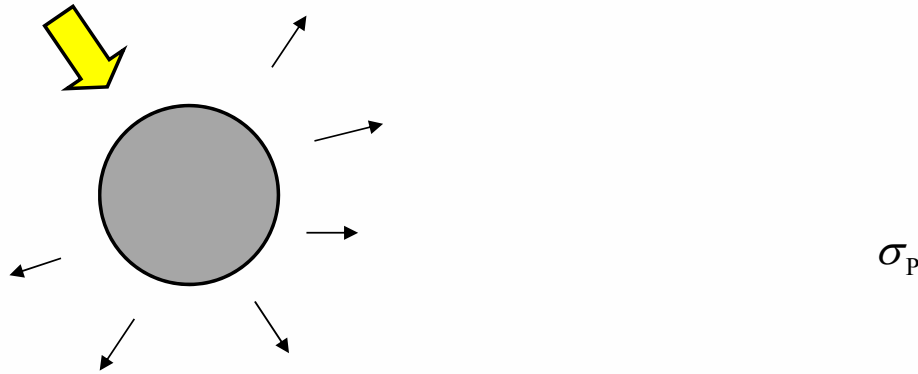
If the particle density is $\rho_p > 1$, the particle is irregular $\chi > 1$, and the sedimentation velocities are $\vec{u}_{s3} = \vec{u}_{s1}$, the aerodynamic diameters are identical $D_{P3,Ae} = D_{P1,Ae}$.

$$D_{P3, Ve} = D_{P1, Ae} \sqrt{\frac{\rho_0}{\rho_p} \chi}$$

Optical Diameter

Optical Diameter

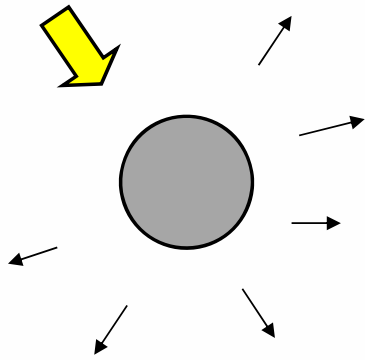
The optical particle diameter is based on the particle scattering σ_p , if the particle is illuminated.



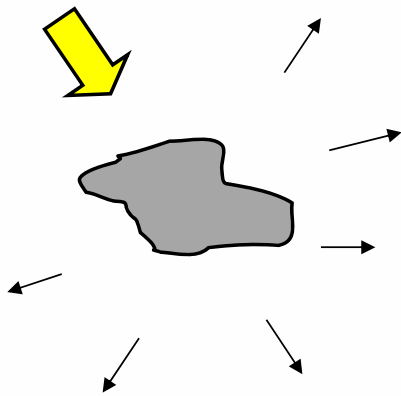
An illuminated spherical particle with a known Stokes diameter (latex particle) and a known refractive index gives a certain particle scattering

The optical diameter of this particle is then calibrated to:

$$D_{P1,Latex} = D_{P1,Opt} = D_{P1,St}$$



An illuminated spherical particle with unknown size and refractive index, but with the same particle scattering $\sigma_{P2} = \sigma_{P1}$ has the same optical diameter $D_{P2,Opt} = D_{P1,Opt}$.



An illuminated irregular particle with unknown shape and refractive index, but with the same particle scattering $\sigma_{P3} = \sigma_{P1}$ has the same optical diameter $D_{P3,Opt} = D_{P1,Opt}$.

Vacuum Aerodynamic Diameter

Vacuum Aerodynamic Diameter

In the free-molecular regime (vacuum), the aerodynamic diameter is called the vacuum aerodynamic diameter

$$\frac{\rho_P \cdot D_P^2 \cdot C_C \cdot \vec{g}}{18\eta} = \frac{\rho_o \cdot D_{P,VAe}^2 \cdot C_C \cdot \vec{g}}{18\eta}$$

The Cunningham Slip Correction factor for the free molecular regime can be simplified to:

$$C_C \approx \frac{\lambda}{D_P}$$

$$\frac{\rho_P \cdot D_P \cdot \vec{g}}{18\eta \cdot \lambda} = \frac{\rho_o \cdot D_{P,VAe} \cdot \vec{g}}{18\eta \cdot \lambda}$$

The volume equivalent diameter can then be determined by:

$$D_{P,Ve} = D_{P,VAe} \frac{\rho_o}{\rho_P}$$

Particle Number Size Distribution

Particle Number Size Distributions

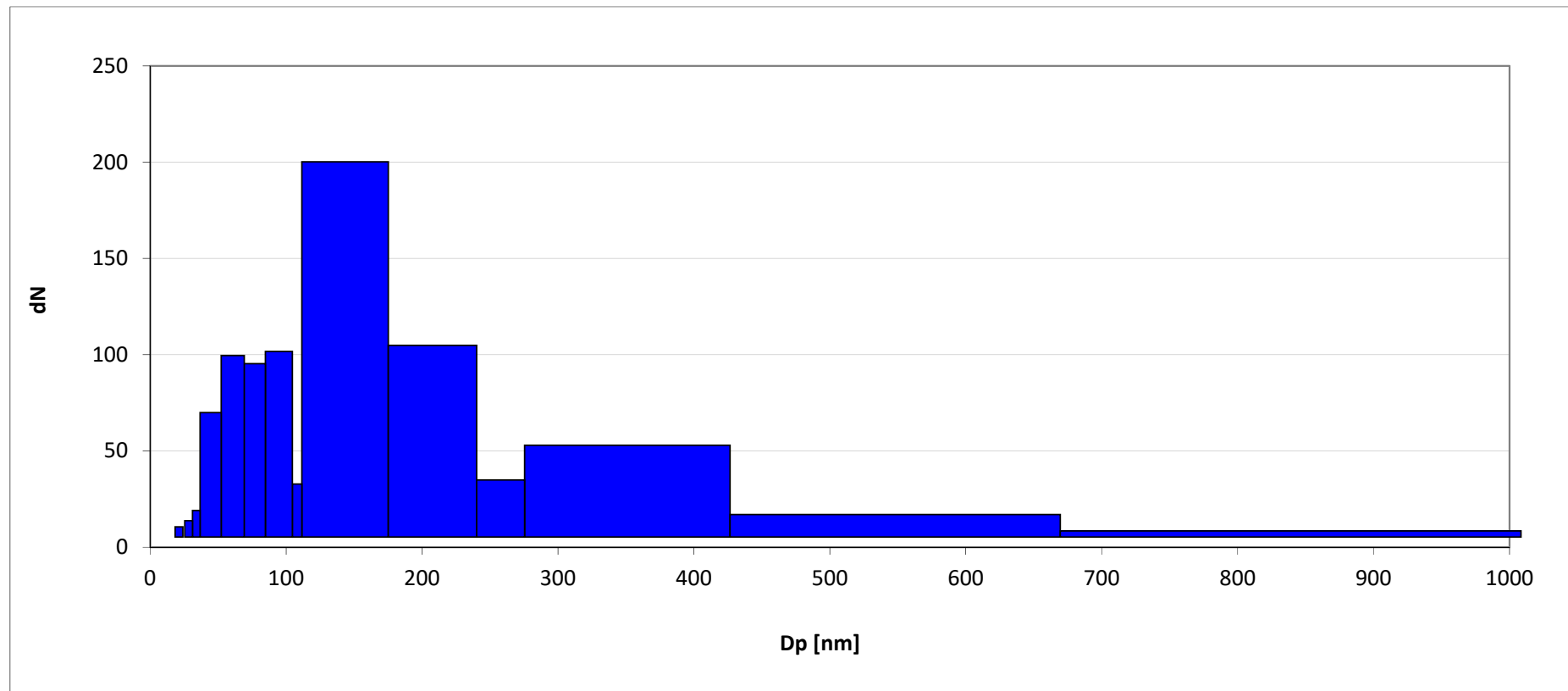
- It would probably be the best to display an aerosol size distribution as a list of concentrations of more than 1000 particle sizes classes.
- However, aerosol instruments normally provide only a limited number of size intervals and concentrations.

Example

An instrument with 10 size classes provides 21 values (upper and lower diameters, and concentrations for each size class).

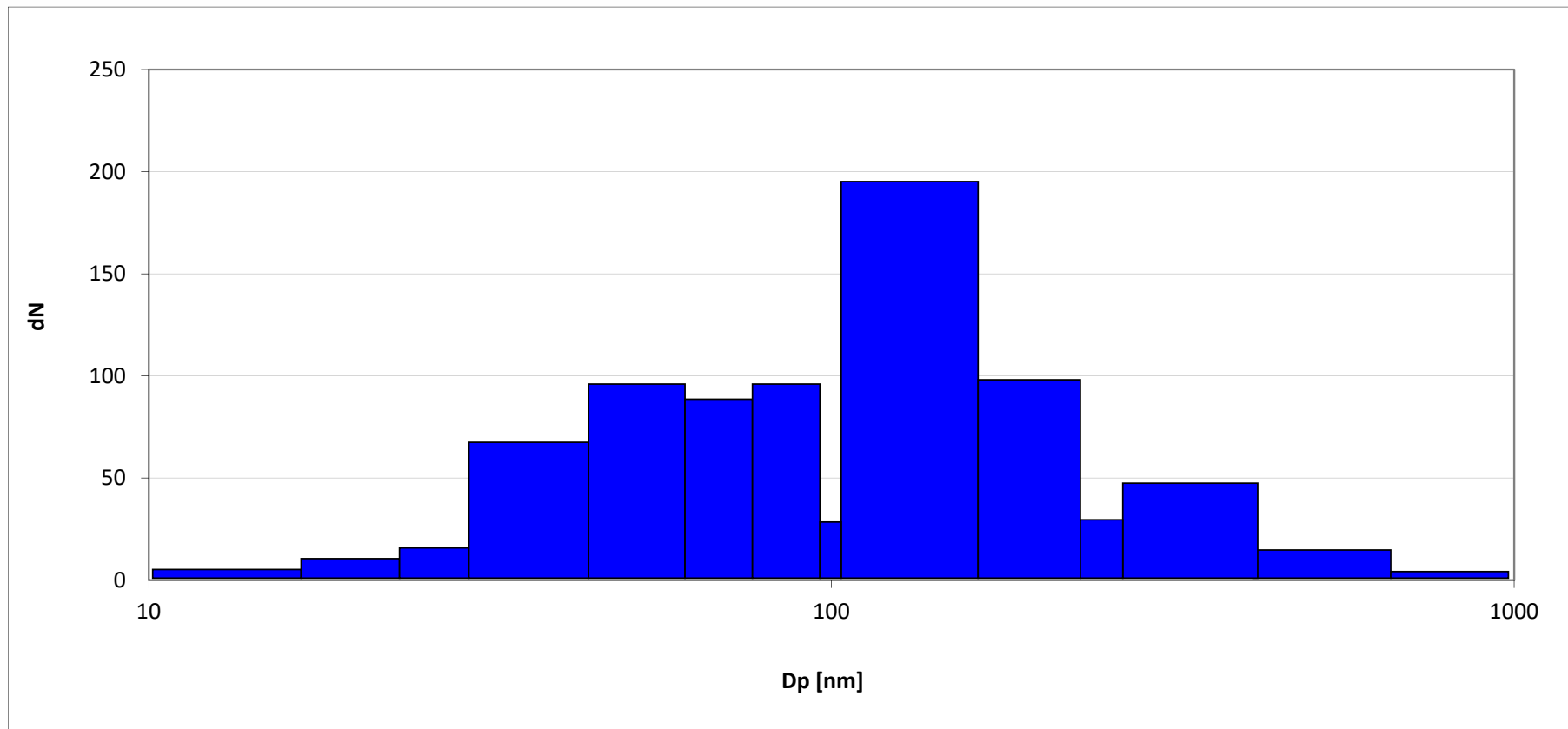
Histogram: Number concentrations N_i in each size interval

It is possible to plot the size distribution as number concentration over particle size. The total number concentration N is the sum of the concentrations N_i of all intervals.

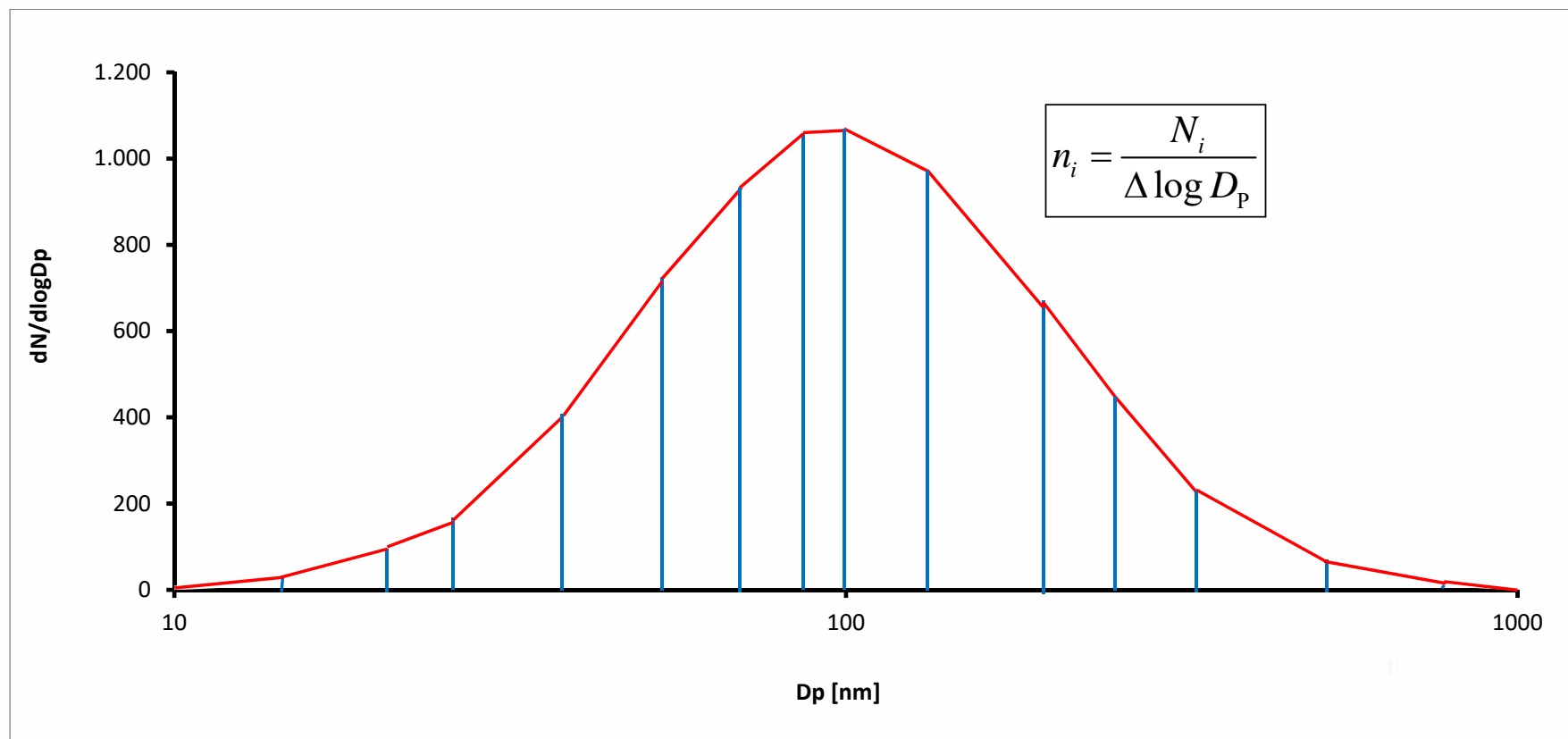


Histogram: Number concentrations N_i in each size class

A change from linear to logarithmic scale is an advantage, if the size range covers more than one order of magnitude (normal case for the atmospheric aerosol)



Number concentration N_i in each size class normalized by the logarithmic width of the size interval N_i is here normalized by the logarithmic width of the size interval.



Lognormal Size Distribution

Log-Normal-Distribution

- The height of humans is normally distributed. Only few people exceed the height of 2.30 m.
- The annually income of people is log-normally distributed. There are people earning more than 1000-times more than normal workers.

Aerosol particle size distributions

- An aerosol particle size distribution can be often described as a log-normal distribution.
- Log-normal-distributions are used to plot particle size distributions, e.g. in the size range from 1-1000 nm.
- Normal- und log-normal-distributions are probability functions.
- The area under a (log)-normal distribution function is unity.
- The log-normal must be multiplied with the total number concentration to describe the particle number distribution.

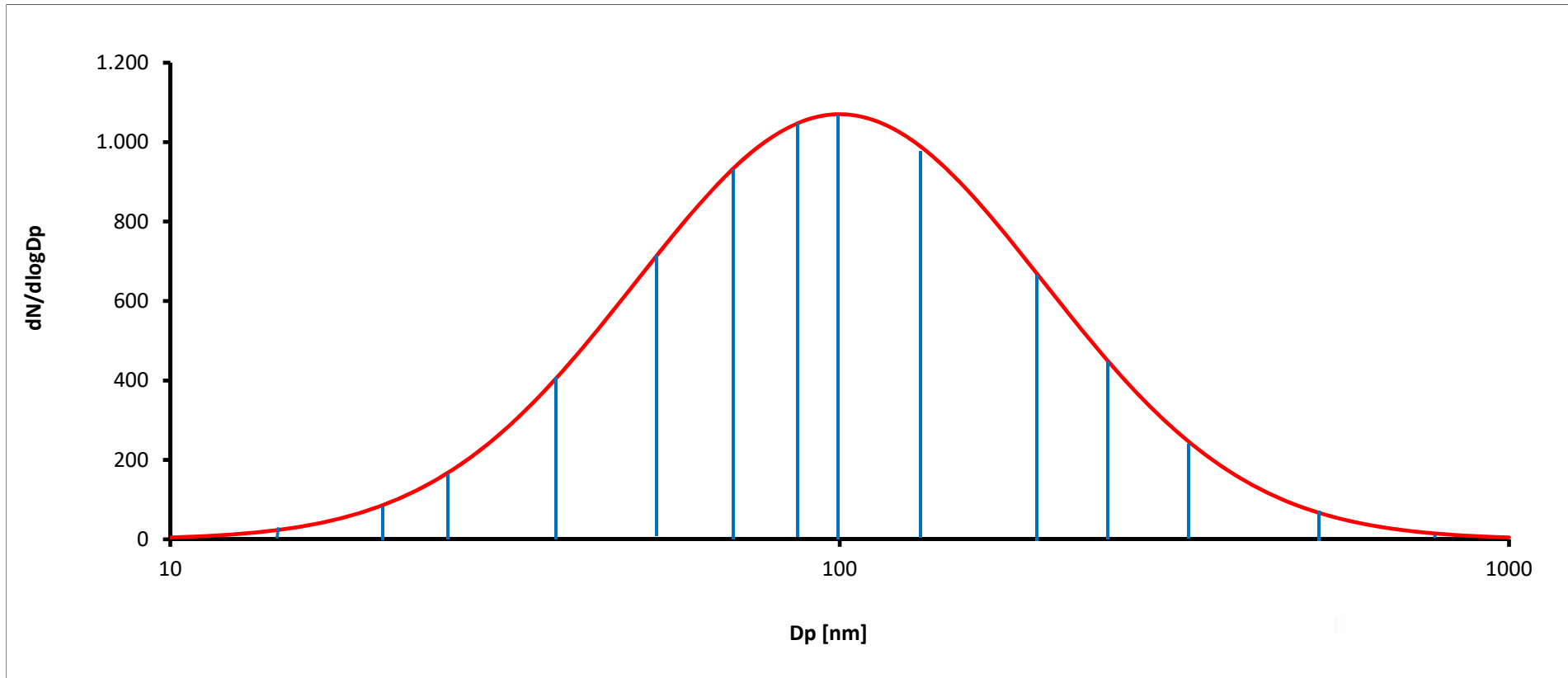
The log-normal-distribution is described by the geometric mean diameter and standard deviation.

$$\frac{df}{d \log D_p} = \frac{N}{\sqrt{2\pi \cdot \log \sigma_g}} \exp \left(-\frac{(\log D_p - \log \overline{D_{p_g}})^2}{2(\log \sigma_g)^2} \right)$$

$$\log \overline{D_{p_g}} = \frac{\sum N_i \cdot \log D_{p_i}}{N}$$

$$\log \sigma_g = \left(\frac{\sum N_i (\log D_p - \log \overline{D_{p_g}})^2}{N - 1} \right)^{1/2}$$

Particle number size distribution: Log-Normal function



Recommendations

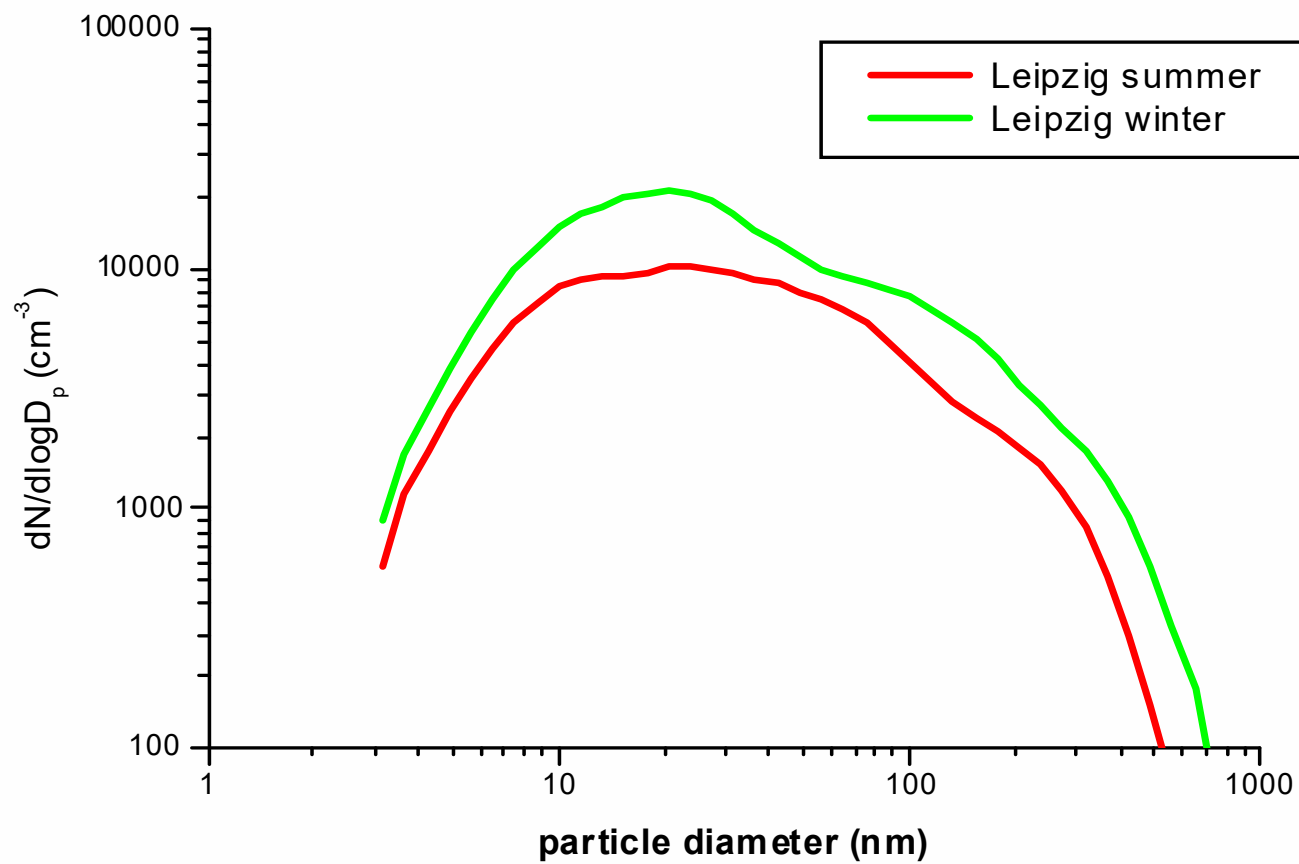
Recommendations

- If the size range (x-axis) covers 1 order of magnitude, use a linear scale.
- If the size range covers 2 or more orders of magnitude, use the logarithmic scale.
- To display a concentration (y-axis), use always the normalized style (e.g. $\Delta N/\Delta D_p$).
- If the concentration range of interest covers 1 order of magnitude, use a linear scale.
- If the concentration range of interest covers 2 or more orders of magnitude, use the logarithmic scale.

Examples of Atmospheric Particle Number Size Distributions

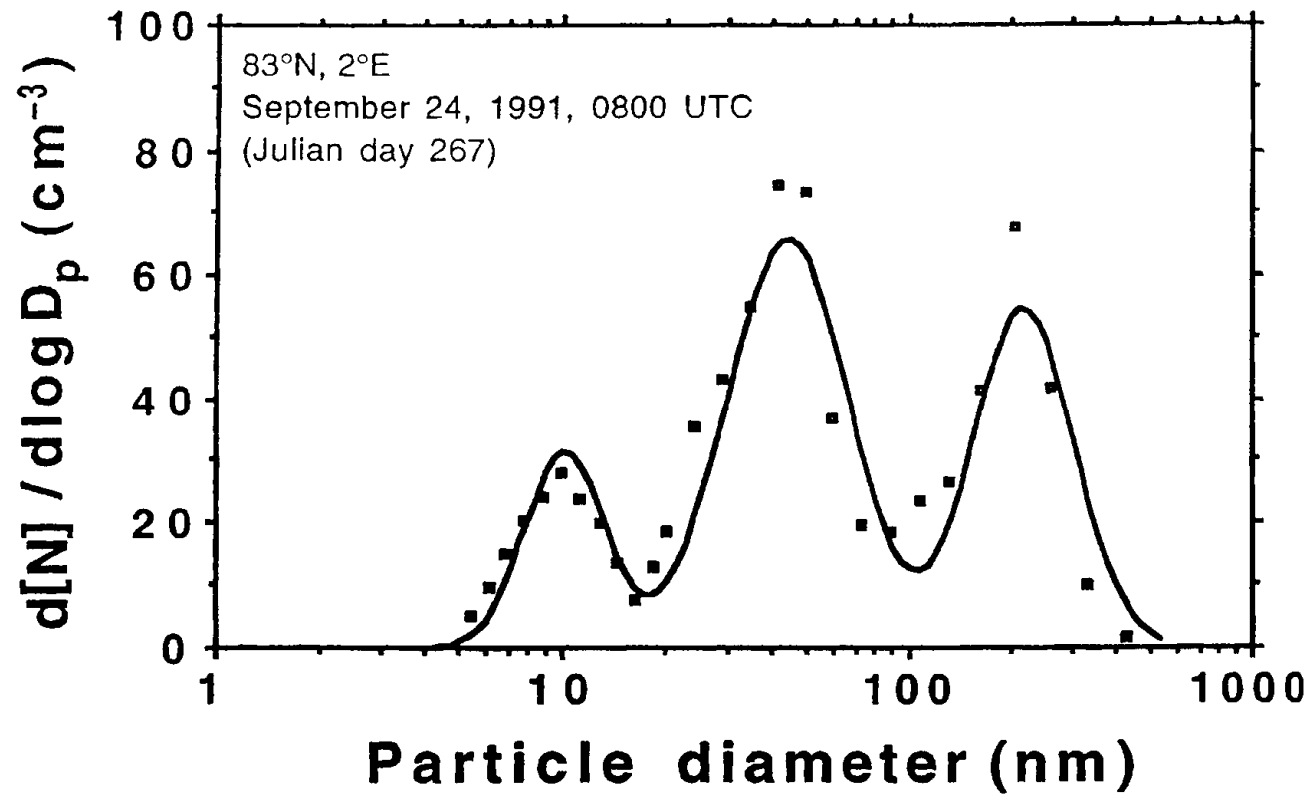
Leipzig

- Mean particle number size distribution measured in Leipzig, Germany, for winter and summer
- Double logarithmic axes



Arctic

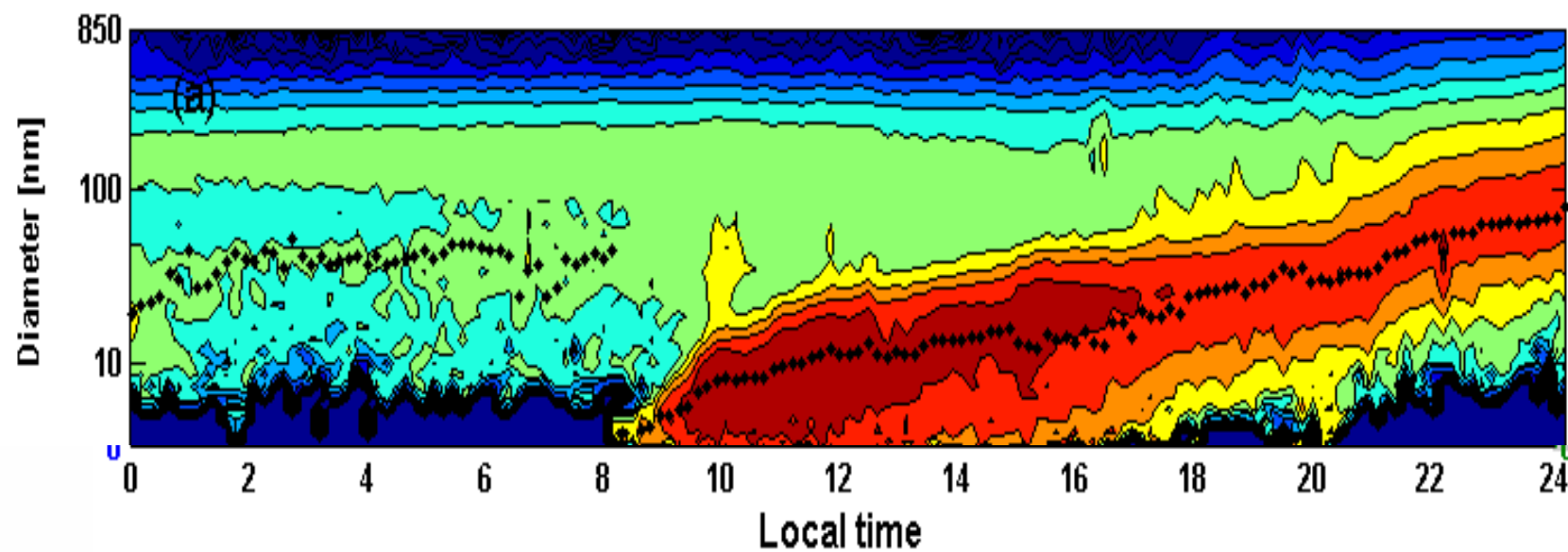
- Particle number size distribution with three log-normal modes measured in the Arctic
- x-axis logarithmic, y-axis linear



Covert, D. S. et al. (1995) *Tellus* **48B**, 197-212.

North China Plain

- Contour plot of the diurnal development of the particle number size distribution after a new particle formation event



Shen, X. J. et al. (2011) ACP **11**, 1565-1580